

A Formal, One-Pot β -Chlorination of Primary Alcohols and Its Utilization in the Transformation of Terpene Feedstock and the Synthesis of a C_2 -Symmetrical Terminal Bis-Epoxide

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Supporting Information

ABSTRACT: A one-pot transformation of alkan-1-ols into 2-chloroalkan-1-ols is described. As a practical application, terpene-derived primary alcohols were converted into semiochemicals such as olfactory lactones (aerangis lactone, whisky lactone, and cognac lactone) and pheromones (cruentol and ferrugineol). Using heptane-1,7-diol as a bifunctional substrate, the corresponding bis-epoxide was synthesized by bidirectional synthesis in good yield and high enantioselectivity.

■ INTRODUCTION

The conversion of biomass into high-value chemicals is a central goal in catalysis.² In recent years, terpenes have been utilized increasingly as feedstock³ for the synthesis of biologically active natural products,⁴ fragrance compounds,⁵ and fine chemicals.⁶ The key methodological challenge in this approach is to achieve site-selective functionalization of the terpene scaffold, thereby allowing for subsequent C-C coupling reactions. To this end, our group has developed protocols for the conversion of terpene aldehydes into the corresponding 1,2-epoxides⁷ as well as an allylic substitution of terpene acetates with vinyl nucleophiles to give 1,4-dienes.8 Taking advantage of both methods, we have assembled a major part of ripostatin B's polyketide backbone from two molecules of geranyl acetate.9

Herein, we describe the development of a telescoped oxidation-chlorination-reduction and its application to the conversion of primary alcohols into enantioenriched 1,2epoxides and a C2-symmetrical terminal bisepoxide via chlorohydrin intermediates. Using terpene-derived alcohols as substrates has allowed us to synthesize a range of semiochemicals while bisepoxides might prove useful in the bidirectional synthesis of polyols.

Halogen-bearing chiral centers 10 constitute a useful handle for subsequent transformations, and halogenation of carbonyl compounds¹¹ provides a highly stereoselective entry into this substrate class. In particular, α -haloaldehydes represent versatile building blocks for heterocyclic scaffolds¹² and for natural product synthesis. 13 For example, the Britton group has demonstrated the broad utility of α -chloroaldehydes in the synthesis of carbohydrates, 14 alkaloids, 15 and acetogenins. 16

The merging of the oxidation and the chlorination step into a single reaction vessel (one-pot reaction)¹⁷ obviates the need to isolate and purify possibly unstable aldehyde intermediates. For this purpose, a mild oxidation protocol was sought that would not interfere with the rather sensitive subsequent organocatalytic chlorination step. In this sense, harsh reaction conditions and reactive stoichiometric byproducts from the terminal oxidant were to be avoided. In 2011, Hoover and Stahl¹⁸ reported a mild protocol for the Cu-catalyzed aerobic oxidation 19 of primary alcohols to aldehydes based on pioneering work by Semmelhack, 20 Markó, 21 Sheldon, 22 and Koskinen. 23 With water as the only stoichiometric byproduct, we envisioned this protocol to be an ideal starting point for telescoping an oxidation with subsequent organocatalytic transformations.²⁴ In a proof-of-concept study, we joined Stahl's oxidation conditions with a DMAP-catalyzed isomerization and an organocatalytic Diels-Alder reaction, 25 while Jang et al. reported a similar oxidation in conjunction with an organocatalytic Michael addition/oxyamination.²⁶

RESULTS AND DISCUSSION

Before we investigated the one-pot reaction, the conditions for the organocatalytic α -chlorination²⁷ were optimized for citronellal (1a) as the substrate. Our previously described procedure is based on the work of Jørgensen^{28,29} and MacMillan^{30,31} and employs catalyst 2³² together with Nchlorosuccinimide (NCS) as an inexpensive chlorine source. The screening of the α -chlorination was carried out using (R)citronellal with 96% ee, and the sensitive α -chloroaldehyde was converted to the corresponding alcohol by in situ reduction with sodium borohydride (Table 1). While the reaction exhibited high diastereoselectivity over a wide temperature range, between 23 and 0 °C the isolation of pure (2S,3R)-3a was hampered by the formation of unidentified side products.

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Table 1. Optimized Organocatalytic α -Chlorination of Citronellal^{α}

 a 30 mol % **2**·TFA, 1.30 equiv of NCS, then 2.50 equiv of NaBH₄, 0 $^{\circ}$ C. b ee of the major diastereomer. c ent-**2** (30 mol %) was used.

This problem was circumvented efficiently by lowering the temperature to -15 °C. Accordingly, the alcohol (2*S*,3*R*)- **3a** was isolated in 73% yield with >99% ee. The *syn/anti* ratio of 98:2 reflects the 96% *ee* of the starting material. Using the enantiomeric MacMillan catalyst *ent*-**2** at -28 °C afforded the *anti*-diastereomer (2*R*,3*R*)-4a in a 98:2 ratio in >99% *ee*. From these results we conclude that the α -chlorination appears to be catalyst-controlled.

We next turned our attention to the possibility of employing alcohols as substrates in a three-step one-pot sequence consisting of oxidation, α -chlorination, and reduction. Exploratory experiments indicated that reagents for the Cu-catalyzed aerobic oxidation do not interfere with chlorination. As shown in Table 2, the reactions using alcohols $\mathbf{5a-c}$ proceeded smoothly and were completely catalyst-controlled. The low *anti/syn* ratio of $\mathbf{4b}$ is a direct consequence of the low *ee* of the

Table 2. Scope of the Telescoped Oxidation—Chlorination—Reduction Sequence a

starting alcohol (5b). Treatment of the chlorohydrins 4a-c with NaOH afforded the corresponding terminal epoxides 6a-c in 93-97% yield.

The epoxide function (a²-synthon) is a useful handle for carbon chain extension. With the optimized procedure in hand, we next demonstrated the utility of monoterpene-derived 1,2-epoxides as chiral building blocks. Apart from the Roche ester,³³ citronellol is one of only a few viable resources for transferring methyl group bearing stereogenic centers from the chiral pool into target molecules and that is readily available in both enantiomeric forms. While citronellol's trisubstituted double bond represents a masked carbonyl group, the primary hydroxyl group can be transformed into a chiral 1,2-epoxide via one-pot oxidation—chlorination.

Thus, the citronellol-derived epoxide **6a** was converted into the main odor component of the African orchids *Aerangis kirkii* and *A. confusa*, (-)-cis-aerangis lactone **8**,³⁴ in just two synthetic steps.^{35–40} As shown in Scheme 1, a Cu-catalyzed

Scheme 1. Synthesis of (-)-cis-Aerangis Lactone

regioselective opening of **6a** with *n*-butyllithium using CuCN (20 mol %) as the catalyst afforded **7a** in 89% yield. Using Marshall's ozonolytic esterification ⁴¹/lactonization ⁴² protocol, (–)-*cis*-aerangis lactone **8** was obtained in 83% yield.

Alternatively, **6a** can be converted into palm weevil pheromones. Epoxide opening with Grignard reagent derived cuprates afforded the alcohols **7b** and **7c** in good yields. Acetylation of the hydroxyl group was followed by an ozonolysis with a reductive workup to give the primary alcohols **9b** and **9c**. Finally, tosylation and global reduction completed the synthesis of cruentol⁴⁴ **10b** and ferrugineol^{45,46} **10c** (Scheme 2).

In a similar vein, 1,2-epoxydiene **6c** was subjected to a CuCN-catalyzed opening with n-BuLi and n-propylmagnesium bromide, respectively to afford the alcohols **11a** (81%) and **11b** (74%). A reductive ozonolysis was followed by an oxidative lactonization and yielded cognac lactone^{47–52} (**12a**, R = n-Bu, 77%) and whisky lactone^{53–57} (**12b**, R = n-Pr, 73%), two major flavor contributing constituents of most alcoholic beverages aged in smoked oak or other wooden casks.⁵⁸ Moreover, avoidance of the isolation of the labile and volatile aldehyde proved time-saving and led to satisfactory overall yields (Scheme 3).

Two-directional chain extension⁵⁹ is a multibond forming approach⁶⁰ to molecules with repeating functional group patterns. While this strategy has been applied to various achiral, prochiral,⁶¹ meso- and C_2 -symmetrical substrates,⁶² examples of regioselective double opening of C_2 -symmetrical terminal bis-epoxides with C-nucleophiles are scarce.⁶³ The currently established synthesis of C_2 -symmetrical bis-epoxides involves peracid-mediated double epoxidation of two terminal

 $[^]a5.0$ mol % CuOTf, 5.0 mol % 2,2′-bipyridine, 5.0 mol % TEMPO, 10 mol % NMI (1-methylimidazole), rt, then 30 mol % 2·TFA, 1.30 equiv of NCS, $-28\,$ °C, then 2.50 equiv of NaBH₄, 0 °C; 22.3 equiv of NaOH, rt.

Scheme 2. Synthesis of Cruentol and Ferrugineol

Scheme 3. Synthesis of Whisky Lactone and Cognac Lactone

alkene moieties⁶⁴ followed by hydrolytic kinetic resolution⁶⁵ of the resulting meso/rac-mixture. 66,67 As a consequence, the theoretical yield is limited to 25% of either of the two enantiomeric C2-symmetrical bis-epoxides and 50% of the mesodiastereomer. To test the hypothesis whether a formal double chlorination of symmetrical diols would serve as an entry into C_2 -symmetrical bis-epoxides, we subjected 1,7-heptanediol (13) to the previously established conditions. Gratifyingly, the (2S,6S)-2,6-dichloroheptane-1,7-diol (14) was obtained as a crystalline solid in good yield and selectivity. The high enantiomeric excess of the C2-symmetrical diastereomer is due to the fact that the minor enantiomer in the first chlorination is preferentially converted to the meso-diastereomer in the second chlorination (Horeau principle).⁶⁸ The minor meso-isomer can be removed efficiently by recrystallization. Treatment of 14 with NaOH in dichloromethane affords the desired bisepoxide 15 in 92% yield and >99% enantiomeric excess (Scheme 4).

CONCLUSION

In summary, we have demonstrated that alkan-1-ols can be converted to the corresponding 2-chloroalkan-1-ols in an efficient one-pot procedure involving a Cu-catalyzed oxidation and an enamine chlorination/reduction. Using terpene derived alcohols as feedstock, several chiral semiochemicals have been synthesized in an efficient manner via the corresponding 1,2-epoxides. Application of this telescoped sequence to bisepoxides in the bidirectional synthesis of more complex targets will be the subject of forthcoming reports from our laboratory.

Scheme 4. Synthesis of Terminal Bis-Epoxide 15

■ EXPERIMENTAL SECTION

(2R,3R)-2-Chloro-3,7-dimethyloct-6-en-1-ol 4a. To a solution of (R)-citronellal 1a (135 mg, 0.88 mmol) in MeCN (1.75 mL) (2S,5R)-2-(tert-butyl)-3,5-dimethylimidazolidin-4-one-TFA-salt (74.8 mg, 0.30 mmol, 0.30 equiv) and NCS (152 mg, 0.65 mmol, 1.30 equiv) were added successively at -28 °C. The reaction was stirred for 8 h at -28 °C and warmed to 0 °C. After the addition of EtOH (1.0 mL) and NaBH₄ (82.8 mg, 2.50 mmol, 2.50 equiv) the reaction was stirred for 30 min at 0 $^{\circ}\text{C}$, quenched by the addition of a saturated aqueous solution of NH₄Cl (2 mL), and extracted with CH₂Cl₂ (3 \times 5 mL). The combined organic layers were dried over MgSO₄ and concentrated under reduced pressure. The crude product was purified by column chromatography on silica (n-pentane/Et₂O = 25:1), and 4a was obtained as a colorless liquid (136 mg, 82%, 98:2 dr, >99:1 er). $[\alpha]_{D}^{20} = +16.7$ (c 0.75, CHCl₃); ¹H NMR (400 MHz, CDCl₃): δ 5.09 (t, J = 7.1 Hz, 1H), 3.98 (ddd, J = 8.3, 5.1, 3.8 Hz, 1H), 3.71-3.85 (m,2H), 2.04-2.13 (m, 2H), 1.88-1.99 (m, 2H), 1.69 (s, 3H), 1.61 (s, 3H), 1.52-1.60 (m, 1H), 1.27-1.35 (m, 1H), 1.03 (d, J = 6.9 Hz, 3H); 13 C NMR (100 MHz, CDCl₃): δ 132.0, 123.9, 71.0, 64.9, 36.3, 32.7, 25.7, 25.3, 17.7, 16.3; IR 3375, 2965, 2928, 2877, 2858, 1455, 1379, 1259, 1199, 1071, 1031, 987, 953, 824, 669 cm⁻¹; HRMS (EI) m/z calcd for C₁₀H₁₉OCl [M]⁺⁺ 190.11189, found 190.11225; calcd for C₁₀H₁₉O³⁷Cl [M]^{•+} 192.10894, found 192.10772; GC (Hydrodexβ-TBDM, 130 °C isotherm, 1.1 mL/min He, 50:1 split) $T_{(2R,3R)} = 25.5$ min, $T_{(2S,3R)} = 26.4$ min, $T_{(2R,3S)} = 29.6$ min, $T_{(2S,3S)} = 31.2$ min, isomeric ratio = 70:1:1:0.

(2S,3S)-2-Chloro-3,7-dimethyloct-6-en-1-ol 4a (One-Pot Procedure). To a solution of (S)-citronellol 5a (3.12 g, 20 mmol) in MeCN (20 mL) were added solutions of CuOTf-4MeCN (377 mg, 0.05 mmol, 0.05 equiv), 2,2'-bipyridine (156 mg, 0.05 mmol, 0.05 equiv), TEMPO (156 mg, 0.05 mmol, 0.05 equiv), and 1methylimidazole (164 mg, 0.10 mmol, 0.10 equiv) each in 20 mL of MeCN. The reaction mixture was degassed for 5 min, the atmosphere in the reaction vessel was replaced by O2, and the reaction mixture was stirred for 2 h at rt. After the reaction was cooled to -28 °C, (2R,5S)-2-(tert-butyl)-3,5-dimethylimidazolidin-4-one-TFA-salt (1.60 g, 6.00 mmol, 0.30 equiv) and NCS (3.47 g, 26.0 mmol, 1.30 equiv) were added successively, and the reaction was stirred 24 h at −28 °C and warmed to 0 °C. After the addition of EtOH (50 mL) and NaBH₄ (189 g, 50.0 mmol, 2.50 equiv) the reaction was stirred for 30 min at 0 °C, quenched by the addition of a saturated aqueous solution of NH₄Cl (100 mL), and extracted with CH₂Cl₂ (3 × 100 mL). The combined organic layers were dried over MgSO₄ and concentrated under reduced pressure. The crude product was purified by column chromatography on silica (n-pentane/Et₂O = 10:1), and 4a was directly submitted to the following reaction conditions.

Alternatively after column chromatography on silica (n-pentane/ $Et_2O = 35:1-25:1-15:1$), 4a can be obtained as a colorless liquid (3.05 g, 79%, 98:2 dr, >99:1 er).

[α]₅₀ = -13.5 (c 1.09, CHCl₃); GC (Hydrodex- β -TBDM, 130 °C isotherm, 1.1 mL/min He, 50:1 split) τ _(2R,3R) = 25.5 min, τ _(2S,3R) = 26.4 min, τ _(2R,3S) = 29.6 min, τ _(2S,3S) = 31.2 min, isomeric ratio = 0:1:1:83.

(25,35,E)-2-Chloro-3,7,11-trimethyldodeca-6,10-dien-1-ol **4b.** The same procedure as above was applied to ($S_{e}E$)-dihydrofarnesol **5b** (449 mg, 2.00 mmol). The crude product was purified by column chromatography on silica (n-pentane/Et₂O = 35:1), and **4b**

was obtained as a colorless liquid (444 mg, 86%, 58:42 dr, 90:10 er). $[\alpha]_D^{20} = -9.64^\circ$ (c 1.03, CHCl₃); ¹H NMR (400 MHz, CDCl₃): δ 5.10 (d, J = 6.0 Hz, 2H), 4.11–4.02 (m, 1H), 3.86–3.79 (m, 2H), 1.87–2.13 (m, 8H), 1.69 (s, 3H), 1.61 (m, 7H), 1.28–1.38 (m, 1H), 1.04 (d, J = 6.8 Hz, 1/2 3H), 0.98 (d, J = 6.7 Hz, 1/2 3H); ¹³C NMR (100 MHz, CDCl₃): δ 135.6, 135.6, 131.4, 131.3, 124.2, 124.2, 123.8, 123.7, 71.1, 70.2, 65.7, 64.9, 39.7 (2C), 36.3, 35.3, 34.2, 32.7, 26.6 (2C), 25.7 (2C), 25.2, 25.1, 17.7 (2C), 16.3, 16.0 (2C), 14.6; IR 3387, 2965, 2926, 2856, 1667, 1452, 1380, 1107, 1074, 1030, 831, 669 cm⁻¹; HRMS (ESI) m/z calcd for C₁₅H₂₉OCl [M+H]⁺ 259.18232, found 259.18244; calcd for C₁₅H₂₉O³⁷Cl [M+H]⁺ 261.17937, found 261.17944; GC (Lipodex E, 130 °C isotherm, 1.1 mL/min He, 20:1 split) T_(2R,3R) = 83.0 min, T_(2S,3R) = 90.1 min, T_(2R,3S) = 97.6 min, T_(2S,3S) = 105.0 min, isomeric ratio = 1:1:20:14.

(2S,3S,E)-2-Chloro-3,7-dimethylocta-5,7-dien-1-ol 4c. The same procedure as above was applied to (S,E)-3,7-dimethylocta-5,7dien-1-ol 5c (425 mg, 2.75 mmol) with a reaction temperature of 50 °C for the oxidation. The crude product was purified by column chromatography on silica (n-pentane/Et₂O = 15:1), and 4c was obtained as a colorless liquid (422 mg, 82%, 95:5 dr, 98:2 er). $\left[\alpha\right]_{D}^{20}$ = -20.8° (c 1.12, CHCl₃); ¹H NMR (400 MHz, CDCl₃): δ 6.19 (d, J =15.4 Hz, 1H), 5.59 (dt, *J* = 15.4, 7.4 Hz, 1H), 4.90 (s, 2H), 3.97 (ddd, *J* = 7.4, 6.2, 3.5 Hz, 1H), 3.74–3.89 (m, 2H), 2.36–2.43 (m, 1H), 2.01– 2.17 (m, 3H), 1.84 (d, J = 0.8 Hz, 3H), 1.03 (d, J = 6.7 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃): δ 141.7, 135.3, 127.2, 115.1, 70.1, 64.9, 37.1, 36.2, 18.6, 16.5; IR 3382, 3080, 3019, 2967, 2937, 1646, 1608, 1454, 1378, 1310, 1136, 1073, 967, 885 cm⁻¹; HRMS (EI) m/z calcd for C₁₀H₁₇OCl [M]^{•+} 188.09624, found 188.09702; calcd for C₁₀H₁₇O³⁷Cl [M]^{•+} 190.09329, found 190.09356; GC (Lipodex E, 100 °C isotherm, 1.1 mL/min He, 20:1 split) $T_{(2R,3R)} = 42.7$ min, $T_{(2S,3R)} = 44.7 \text{ min, } T_{(2R,3S)} = 63.5 \text{ min, } T_{(2S,3S)} = 71.7 \text{ min, isomeric ratio}$ = 1:2:357:20.

(S,E)-3,7-Dimethylocta-5,7-dien-1-ol 5c. To a solution of SeO₂ (140 mg, 1.26 mmol, 0.05 equiv) and salicylic acid (348 mg, 2.52 mmol, 0.10 equiv) in CH₂Cl₂ (12 mL) at rt, t-BuOOH (70% in H₂O, 15.6 mL, 114 mmol, 4.50 equiv) was added slowly. After 15 min (S)citronellol 5a (5.00 g, 25.2 mmol) was added and stirred 48 h at rt. SeO₂ (140 mg, 1.26 mmol, 0.05 equiv) was added and stirred for another 72 h at rt. The reaction was quenched by the addition of MeOH (25 mL) and a saturated aqueous solution of Na₂S₂O₃ (40 mL). After 1 h of stirring at rt the reaction was filtered and extracted with Et₂O (3 \times 50 mL), the combined organic layers were washed with an aqueous solution of NaOH (1 M, 200 mL), and the basic aqueous layer was washed with Et₂O (1 × 100 mL). The combined organic layers were washed with a saturated aqueous solution of NH₄Cl (1 × 150 mL), dried over Na₂SO₄, and concentrated under reduced pressure. The crude product was solved in dry MeOH (25 mL) and cooled to 0 °C, and NaBH₄ (2.39 g, 63.0 mmol, 2.5 equiv) was added successively. After stirring for 1 h at 0 °C the reaction was quenched by the addition of a saturated aqueous solution of NH₄Cl (25 mL) and extracted with CH_2Cl_2 (3 × 50 mL). The combined organic layers were dried over Na2SO4 and concentrated under reduced pressure. The crude product was purified by column chromatography on silica (n-pentane/EtOAc = 3:1-2:1-1:1-0:1), and (S,E)-2,6-dimethyloct-2-en-1,8-diol was obtained as a colorless liquid (3.69 g, 66%).

To a solution of (S,E)-2,6-dimethyloct-2-en-1,8-diol (2.88 g, 16.7 mmol) in dry CH₂Cl₂ (17 mL) at 0 °C were added Et₃N (6.13 mL, 44.2 mmol, 2.64 equiv), DMAP (205 mg, 1.67 mmol, 0.10 equiv), and Ac₂O (3.80 mL, 40.18 mmol, 2.40 equiv) and stirred for 1.5 h at 0 °C. The reaction was quenched by the addition of a saturated aqueous solution of NaHCO₃ (20 mL) and extracted with CH₂Cl₂ (3 × 50 mL), and the combined organic layers dried over MgSO₄ and concentrated under reduced pressure. The crude product was purified by column chromatography on silica (n-pentane/EtOAc = 30:1), and (S,E)-2,6-dimethyloct-2-en-1,8-diyldiacetate was obtained as a colorless liquid (4.20 g, 98%).

To a solution of (S,E)-2,6-dimethyloct-2-en-1,8-diyldiacetate (1.28 g, 5.00 mmol) in dry THF (15 mL) at rt were added ZnCl₂ (682 mg, 5.00 mmol, 1.00 equiv) and Pd(PPh₃)₄ (57.8 mg, 0.05 mmol, 0.01

equiv). Over 30 min AllylMgCl (1.7 M in Et₂O, 5.88 mL, 10.0 mmol, 2.00 equiv) was added, and the reaction was stirred 1.5 h at rt. The reaction was quenched by the addition of a saturated aqueous solution of NH₄Cl-Lsg. (10 mL) and H₂O (20 mL) and extracted with CH₂Cl₂ $(3 \times 35 \text{ mL})$, and the combined organic layers were dried over MgSO₄ and concentrated under reduced pressure. The crude product was solved in MeOH (5 mL) at rt, and K₂CO₃ (1.83 g, 12.5 mmol, 2.50 equiv) was added and stirred 17 h at rt. The reaction was quenched by the addition of a saturated aqueous solution of NH₄Cl-Lsg. (2.5 mL) and extracted with Et₂O (3 × 15 mL), and the combined organic layers were dried over $MgSO_4$ and concentrated under reduced pressure. The crude product was purified by column chromatography on silica (n-pentane/EtOAc = 25:1), and 5c was obtained as a colorless liquid (612 mg, 80%). (S,E)-2,6-Dimethyloct-2-en-1,8-diol: $[\alpha]_{D}^{20} = -4.25^{\circ}$ (c 0.91, CHCl₃); ¹H NMR (400 MHz, CDCl₃): δ 5.35 (td, J = 7.2, 1.1 Hz, 1H), 3.92 (s, 2H), 3.54-3.65 (m, 2H), 2.72 (br. s., 2H)2H), 1.94-2.09 (m, 2H), 1.62 (s, 3H), 1.51-1.59 (m, 2H), 1.29-1.38 (m, 2H), 1.12-1.23 (m, 1H), 0.87 (d, J = 6.5 Hz, 3H); 13 C NMR (100 MHz, CDCl₂): δ 134.5, 126.0, 68.5, 60.6, 39.6, 36.6, 28.8, 24.9, 19.4, 13.5; IR 3320, 2923, 1455, 1377, 1219, 1057, 1011, 848, 667 cm⁻¹; HRMS (ESI) m/z calcd for $C_{10}H_{17}$ [M + H - $2H_2O$]⁺ 137.13248, found 137.13219; calcd for $C_{10}H_{19}O$ [M + H - H_2O] + 155.14304, found 155.14290, calcd for $C_{10}H_{21}O_2$ [M + H]⁺ 173.15361, found 173.15347. (*S,E*)-2,6-Dimethyloct-2-en-1,8-diyldiacetate: $[\alpha]_{D}^{20} =$ -2.01° (c 1.89, CHCl₃); ¹H NMR (400 MHz, CDCl₃): δ 5.38 (t, I = 7.1 Hz, 1H), 4.38 (s, 2H), 3.98-4.08 (m, 2H), 1.92-2.07 (m, 8H), 1.56-1.66 (m, 4H), 1.45-1.54 (m, 1H), 1.29-1.42 (m, 2H), 1.12-1.21 (m, 2H), 0.86 (d, J = 7.3 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃): δ 170.8, 170.6, 129.9, 129.4, 70.0, 62.6, 36.2, 35.2, 29.3, 24.9, 20.8, 20.7, 19.2, 13.7; IR 2959, 2927, 2873, 1743, 1458, 1369, 1241, 1147, 1024, 957, 850, 639, 608 cm $^{-1}$; HRMS (ESI) m/z calcd for $C_{14}H_{25}O_4$ [M + H]⁺ 257.17474, found 257.17506; calcd for C₁₄H₂₅O₄Na [M + Na]⁺ 279.15668, found 279.15701. **5c**: $[\alpha]_D^{20} = -8.62^{\circ}$ (c 1.39, CHCl₃); ¹H NMR (400 MHz, CDCl₃): δ 6.13 (d, J = 15.6 Hz, 1H), 5.59-5.66 (m, 1H), 4.86 (s, 2H), 3.60-3.72 (m, 2H), 2.13 (dt, J =13.9, 7.0 Hz, 1H), 1.95-2.02 (m, 1H), 1.82-1.84 (m, 3H), 1.79 (br. s., 1H), 1.59-1.73 (m, 2H), 1.35-1.42 (m, 1H), 0.91 (d, J = 6.5 Hz, 3H); 13 C NMR (100 MHz, CDCl₃): δ 142.0, 134.3, 128.8, 114.3, 60.9, 40.2, 39.3, 29.9, 19.5, 18.6; IR 3337, 3080, 2925, 1741, 1644, 1608, 1455, 1376, 1242, 1055, 1003, 966, 882 cm⁻¹; HRMS m/z calcd for $C_{10}H_{17}$ [M + H - H_2O]⁺ 137.13248, found 137.13227.

(R)-2-((S)-6-Methylhept-5-en-2-yl)oxirane 6a. To a solution of 4a (3.23 g, 17.0 mmol) in MeCN (20 mL) was added a solution of NaOH (17.8 g, 0.45 mmol, 22.3 equiv) in EtOH (24 mL) and H₂O (50 mL). After the reaction was stirred for 2 h at rt, n-pentane (50 mL) was added, the layers were separated, and the aqueous layer was extracted with *n*-pentane (3 \times 50 mL). The combined organic layers were dried over Na₂SO₄ and concentrated under reduced pressure. The crude product was purified by column chromatography on silica (n-pentane/Et₂O = 20:1), and **6a** obtained as a colorless liquid (2.53 g, 97%, 94:6 dr). $[\alpha]_D^{20} = -3.97$ (c 0.92, CHCl₃); ¹H NMR (400 MHz, CDCl₃): δ 5.07 (td, J = 6.4, 1.3 Hz, 1H), 2.71–2.76 (m, 1H), 2.65– 2.70 (m, 1H), 2.51 (dd, *J* = 5.0, 2.7 Hz, 1H), 1.95–2.08 (m, 2H), 1.67 (s, 3H), 1.59 (s, 3H), 1.38-1.47 (m, 1H), 1.24-1.33 (m, 2H), 1.02 (d, I = 6.5 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃): δ 131.6, 124.2, 56.9, 46.8, 35.7, 33.6, 25.6, 25.5, 17.6 16.9; IR 3043, 2966, 2919, 2856, 1483, 1455, 1376, 1257, 1114, 1088, 929, 916, 892, 856, 823 cm⁻¹; HRMS (ESI) calcd for C₁₀H₁₉O [M+H]⁺ 155.14304, found 155.14289.

(*R*)-2-((*S*,*E*)-6,10-Dimethylundeca-5,9-dien-2-yl)oxirane **6b.** The same procedure as above was applied to (2*S*,3*S*,*E*)-4b (597 mg, 2.31 mmol). The crude product was purified by column chromatography on silica (*n*-pentane/Et₂O = 100:1), and **6b** was obtained as a colorless liquid (480 mg, 93%, 56:44 dr). ¹H NMR (400 MHz, CDCl₃): δ 5.05-5.16 (m, 2H), 2.66-2.77 (m, 2H), 2.46-2.53 (m, 1H), 2.02-2.10 (m, 4H), 1.96-2.01 (m, 2H), 1.68 (d, *J* = 0.9 Hz, 3H), 1.59-1.62 (m, 6H), 1.43 (s, 4H), 1.26-1.38 (m, 2H), 1.03 (d, *J* = 6.4 Hz, 1/2 3H), 0.93 (d, *J* = 6.5 Hz, 1/2 3H); ¹³C NMR (100 MHz, CDCl₃): δ 135.2, 135.1, 131.3, 131.2, 124.3, 124.3, 124.2, 124.1, 56.9, 56.9, 46.8, 45.5, 39.7, 39.7, 35.7, 35.6, 34.5, 33.6, 26.9 (2C), 26.7, 26.6,

25.6, 25.3, 25.2, 17.6, 17.0, 15.9, 15.9, 15.4; HRMS (ESI) calcd for $C_{15}H_{27}O [M + H]^+$ 223.20578, found 223.20564.

(*R*)-2-((*S*,*E*)-6-Methylhepta-4,6-dien-2-yl)oxirane 6c. The same procedure as above was applied to (2*S*,3*S*,*E*)-4c (422 mg, 2.24 mmol). The crude product was purified by column chromatography on silica (*n*-pentane/Et₂O = 20:1), and 6c was obtained as a colorless liquid (317 mg, 93%, 96:4 dr). [α]_D²⁰ = +15.8° (*c* 1.25, CHCl₃); ¹H NMR (400 MHz, CDCl₃): δ 6.18 (d, *J* = 15.4 Hz, 1H), 5.63 (dt, *J* = 15.6, 7.4 Hz, 1H), 4.89 (s, 2H), 2.73–2.77 (m, 2H), 2.54 (t, *J* = 4.4 Hz, 1H), 2.21–2.28 (m, *J* = 7.5, 6.5 Hz, 1H), 2.06–2.13 (m, 1H), 1.84 (s, 3H), 1.39–1.48 (m, 1H), 1.04 (d, *J* = 6.7 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃): δ 141.8, 134.6, 127.8, 114.8, 56.6, 46.7, 36.9, 36.5, 18.7, 16.7; IR: 2968, 2921, 1608, 1456, 1375, 966, 883, 756 cm⁻¹; HRMS *m/z* calcd for C₁₀H₁₇O [M + H]⁺ 153.12739, found 153.12722.

(65,75)-7,11-Dimethyldodec-10-en-6-ol 7a. To a solution of CuCN (254 mg, 2.84 mmol, 0.20 equiv) in THF (25 mL) at -78 °C were added 6a (2.20 g, 14.2 mmol) and, after 20 min of stirring, dropwise n-BuLi (1.6 M in n-hexane, 28 mL, 2.00 equiv). The reaction was stirred for 7 h at -78 °C, warmed to 0 °C, quenched by the addition of a saturated aqueous solution of NH₄Cl (25 mL), and extracted with EtOAc (3 × 25 mL). The combined organic layers were dried over MgSO₄ and concentrated under reduced pressure. The crude product was purified by column chromatography on silica (npentane/EtOAc = 220:1-150:1-100:1), and 7a was obtained as a colorless liquid (2.66 g, 89%, 92:8 dr). $[\alpha]_D^{20} = -5.9$ (c 1.04, CHCl₃); 1 H NMR (400 MHz, CDCl₃): δ 5.07–5.12 (m, 1H), 3.47–3.51 (m, 1H), 1.91-2.08 (m, 2H), 1.68 (s, 3H), 1.60 (s, 3H), 1.37-1.53 (m, 6H), 1.25–1.36 (m, 5H), 1.16–1.25 (m, 1H), 0.85–0.92 (m, 6H); ¹³C NMR (100 MHz, CDCl₃): δ 131.3, 124.6, 75.0, 37.7, 34.4, 33.4, 31.9, 25.9, 25.7, 25.6, 22.6, 17.6, 14.0, 13.5; IR 3377, 2959, 2928, 2857, 1458, 1378, 1118, 1082, 1013, 983, 943, 828 cm⁻¹; HRMS (ESI) calcd for C₁₄H₂₉O [M+H]⁺ 213.22129, found 213.22134.

(45,55)-5,9-Dimethyldec-8-en-4-ol 7b. The same procedure as above was applied to 6a (849 mg, 5.50 mmol), using EtMgBr (1.0 M in THF, 16.5 mL, 3.00 equiv). The crude product was purified by column chromatography on silica (n-pentane/EtOAc = 250:1), and 7b was obtained as a colorless liquid (803 mg, 79%, 93:7 dr). [α] $_{0}^{20}$ = -22.7 (c 1.33, CHCl $_{3}$); $_{1}^{1}$ H NMR (400 MHz, CDCl $_{3}$): $_{2}^{1}$ S 5.10 (t, $_{3}^{1}$ J = 7.1 Hz, 1H), 3.51 (dt, $_{3}^{1}$ J = 7.5, 4.0 Hz, 1H), 1.91–2.09 (m, 2H), 1.68 (s, 3H), 1.60 (s, 3H), 1.27–1.53 (m, 7H), 1.16–1.25 (m, 1H), 0.93 (t, $_{3}^{1}$ J = 7.1 Hz, 3H), 0.87 (d, $_{3}^{1}$ J = 6.7 Hz, 3H); $_{3}^{1}$ C NMR (100 MHz, CDCl $_{3}$): $_{3}^{2}$ S 131.3, 124.6, 74.8, 37.7, 36.6, 33.4, 25.7, 25.6, 19.4, 17.6, 14.1, 13.5; IR 2280, 2960, 2938, 2872, 1457, 1378, 1113, 974 cm $_{3}^{-1}$; HRMS (ESI) calcd for C $_{12}$ H $_{25}$ O [M + H] $_{3}^{+}$ 185.18999, found 185.18994.

(55,65)-6,10-Dimethylundec-9-en-5-ol 7c. The same procedure as above was applied to 6a (849 mg, 5.50 mmol), using *n*PrMgCl (2.0 M in Et₂O, 5.5 mL, 2.00 equiv). The crude product was purified by column chromatography on silica (*n*-pentane/EtOAc = 250:1), and 7c was obtained as a colorless liquid (803 mg, 74%, 93:7 dr). $[\alpha]_{0}^{20} = -23.1$ (*c* 1.18, CHCl₃); ¹H NMR (400 MHz, CDCl₃): δ 5.10 (tt, J = 7.1, 1.3 Hz, 1H), 3.49 (dt, J = 7.6, 4.0 Hz, 1H), 1.91–2.09 (m, 2H), 1.68 (d, J = 0.7 Hz, 3H), 1.60 (s, 3H), 1.16–1.53 (m, 11H), 0.91 (t, J = 7.1 Hz, 3), 0.87 (d, J = 6.7 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃): δ 131.3, 124.6, 75.0, 37.7, 34.1, 33.4, 28.4, 25.7, 25.6, 22.8, 17.6, 14.0, 13.5; IR 3380, 2959, 2928, 2859, 1457, 1378, 1116, 1079, 978 cm⁻¹; HRMS (ESI) calcd for C₁₃H₂₇O [M + H]⁺ 199.20564, found 199.20565.

(55,65)-5-Methyl-6-pentyltetrahydro-2*H*-pyran-2-one 8, (–)-cis-Aerangis Lactone. A solution of 7a (106 mg, 0.50 mmol) and NaOH (200 mg, 5.00 mmol, 10.0 equiv) in MeOH–CH₂Cl₂ (1:1, 5 mL) was cooled to -78 °C, and ozone enriched O₂ was bubbled through the solution until the color changed from red to yellow and finally to blue with a yellowish precipitate. Air was bubbled through the solution until the blue color disappeared. The precipitate was dissolved by the addition of water (5 mL). After extraction with CH₂Cl₂ (3 × 10 mL) the combined organic layers were dried over MgSO₄ and concentrated under reduced pressure. The crude product was purified by column chromatography on silica (n-pentane/Et₂O = 5:1), and 8 was obtained as a colorless liquid (75.6 mg, 83%, 95:5 dr).

[α]_D²⁰ = -57.8 (c 1.01, CHCl₃); ¹H NMR (400 MHz, CDCl₃): δ 4.25 (ddd, J = 8.6, 4.4, 3.1 Hz, 1H), 2.50 (t, J = 7.3 Hz, 2H), 1.94–2.05 (m, 2H), 1.58–1.68 (m, 2H), 1.41–1.55 (m, 2H), 1.22–1.34 (m, 5H), 0.93 (d, J = 6.9, 3H), 0.86 (t, J = 6.7, 3H); ¹³C NMR (100 MHz, CDCl₃): δ 172.0, 82.9, 31.8, 31.5, 29.2, 26.6, 25.9, 25.1, 22.4, 13.9, 12.3; IR 3465, 2934, 2872, 1735, 1462, 1380, 1344, 1244, 1204, 1123, 1097, 1070, 994, 909, 732 cm⁻¹; HRMS (ESI) calcd for C₁₁H₂₁O₂ [M + H]⁺ 185.15361, found 185.15422.

(4S,5S)-8-Hydroxy-5-methyloctan-4-yl Acetate 9b. To a solution of 7b (777 mg, 4.21 mmol) in dry CH2Cl2 (21 mL) at 0 °C were added Et₃N (1.54 mL, 11.2 mmol, 2.64 equiv), DMAP (102 mg, 0.84 mmol, 0.20 equiv), Ac₂O (0.96 mL, 10.1 mmol, 2.40 equiv) and stirred for 2.5 h at 0 °C. The reaction was quenched by the addition of a saturated aqueous solution of NaHCO₃ (10 mL) and extracted with CH_2Cl_2 (3 $\stackrel{-}{\times}$ 20 mL), and the combined organic layers were dried over MgSO₄ and concentrated under reduced pressure. The crude product was purified by column chromatography on silica $(n\text{-pentane/Et}_2O = 25:1)$ and (4S,5S)-5,9-dimethyldec-8-en-4-yl acetate was obtained as a colorless liquid (914 mg, 96%, 97:3 dr). A solution of (4S,5S)-5,9-dimethyldec-8-en-4-yl acetate (869 g, 3.84 mmol) in MeOH (20 mL) was cooled to -78 °C, ozone enriched O_2 was bubbled through the solution until it turned blue, followed by stirring for 5 min, and air was bubbled through the solution until the blue color disappeared. The reaction was quenched by the addition of NaBH₄ (348 mg, 9.20 mmol, 2.40 equiv), stirred for 1 h while warming to rt, and extracted with CH_2Cl_2 (3 × 25 mL), and the combined organic layers were dried over MgSO4 and concentrated under reduced pressure. The crude product was purified by column chromatography on silica (n-pentane/ $Et_2O = 2:1-1:1$), and 9b was obtained as a colorless liquid (632 mg, 80%, 96:4 dr). (4S,5S)-5,9-Dimethyldec-8-en-4-yl acetate: $[\alpha]_D^{20} = -21.2$ (c 1.46, CHCl₃); ¹H NMR (400 MHz, CDCl₃): δ 5.06 (tt, J = 6.5, 1.5 Hz, 1H), 4.86 (dt, J = 8.7, 4.3 Hz, 1H), 1.89-2.06 (m, 5H), 1.57-1.69 (m, 7H), 1.53 (dd, I = 9.6, 5.0 Hz, 1H), 1.21–1.48 (m, 4H), 1.09–1.17 (m, 1H), 0.87–0.92 (m, 6H); 13 C NMR (100 MHz, CDCl₃): δ 170.8, 131.4, 124.4, 76.8, 35.8, 33.5, 33.0, 25.6, 25.6, 21.0, 18.9, 17.6, 14.3, 13.9; IR 2962, 2930, 2874, 1736, 1456, 1375, 1241, 1019 cm⁻¹; HRMS (ESI) calcd for $C_{14}H_{27}O_2 [M + H]^+$ 227.20056, found 227.20060. 9b: $[\alpha]_D^{20} = -27.4$ (c 1.52, CHCl₃); ¹H NMR (400 MHz, CDCl₃): δ 4.85–4.90 (m, 1H), 3.60 (t, J = 6.5 Hz, 2H), 2.03 (d, J = 0.8 Hz, 3H), 1.11 - 1.69 (m, 10H),0.86-0.93 (m, 6H); ¹³C NMR (100 MHz, CDCl₃): δ 171.1, 76.7, 63.0, 36.2, 33.5, 30.4, 29.1, 21.1, 19.0, 14.4, 13.9; IR 3397, 2960, 2937, 2874, 1733, 1458, 1373, 1247, 1056, 1022, 756, 406 cm⁻¹; HRMS (ESI) calcd for $C_{11}H_{23}O_3$ [M + H] 203.16417, found 203.16423.

(45,55)-1-Hydroxy-4-methylnonan-5-yl Acetate 9c. The same procedure as above was applied to 7c (771 mg, 3.89 mmol). The crude products were purified by column chromatography on silica as above. (5S,6S)-6,10-Dimethylundec-9-en-5-yl acetate (916 mg, 98%, 93:7 dr) and 9c (761 mg, 97%, 97:3 dr) were obtained as colorless liquids. (5S,6S)-6,10-Dimethylundec-9-en-5-yl acetate: $[\alpha]_D^{20} = -16.6$ (c 1.12, CHCl₃); ¹H NMR (400 MHz, CDCl₃): δ 5.04–5.09 (m, 1H), 4.84 (dt, J = 8.5, 4.4 Hz, 1H), 1.89-2.06 (m, 5H), 1.58-1.68 (m, 7H),1.43-1.57 (m, 2H), 1.17-1.42 (m, 5H), 1.08-1.17 (m, 1H), 0.84-0.91 (m, 6H); 13 C NMR (100 MHz, CDCl₃): δ 170.8, 131.4, 124.4, 77.0, 35.7, 33.0, 31.0, 27.9, 25.6, 25.6, 22.6, 21.1, 17.6, 14.2, 13.9; IR 2960, 2929, 2860, 1736, 1456, 1375, 1241, 1018 cm⁻¹; HRMS (ESI) calcd for $C_{15}H_{29}O_2$ [M + H]⁺ 241.21621, found 241.21622. 9c: $[\alpha]_D^{20}$ = -19.4 (c 1.36, CHCl₃); ¹H NMR (400 MHz, CDCl₃): δ 4.86 (dt, J =8.7, 4.3 Hz, 1H), 3.61 (t, J = 6.5 Hz, 2H), 2.04 (s, 3H), 1.12–1.68 (m, 12H), 0.85–0.92 (m, 6H); 13 C NMR (100 MHz, CDCl₃): δ 171.1, 77.0, 76.7, 63.0, 36.2, 31.0, 30.5, 29.1, 27.9, 22.6, 21.1, 14.3, 13.9; IR 3397, 2957, 2936, 2872, 1735, 1716, 1458, 1373, 1242, 1057, 1021, 406 cm $^{-1}$; HRMS (ESI) calcd for $C_{12}H_{25}O_3$ [M + H] $^+$ 217.17982, found 217.18003

(45,55)-5-Methyloctan-4-ol 10b, Cruentol. To a solution of 9b (405 mg, 2.00 mmol) in dry CH_2Cl_2 (1.4 mL) at 0 °C were added DMAP (48.9 mg, 0.40 mmol, 0.20 equiv), Et_3N (360 μ L, 2.60 mmol, 1.30 equiv), and TsCl (458 mg, 2.40 mmol, 1.20 equiv) and stirred for 4 h while warming to rt. The reaction was quenched by the addition of H_2O (2 mL) and extracted with EtOAc (3 × 5 mL), and the combined

organic layers were dried over Na2SO4 and concentrated under reduced pressure. The crude product was purified by column chromatography on silica (n-pentane/Et₂O = 5:1) and (4S,5S)-5methyl-8-(tosyloxy)octan-4-yl acetate obtained as a yellowish liquid (639 mg, 90%, 94:6 dr). To a solution of LiAlH₄ (171 mg, 4.53 mmol, 3.00 equiv) in dry THF (4.53 mL) at 0 °C a solution of (4S,5S)-5methyl-8-(tosyloxy)octan-4-yl acetate (538 mg, 1.51 mmol) in dry THF (2.16 mL) was added and heated to 75 °C for 5 h. After cooling to 0 °C the reaction was quenched by the addition of MeOH (5 mL) and a saturated aqueous solution of NaK-tatrate (10 mL) and extracted with EtOAc (3 × 25 mL), and the combined organic layers were dried over Na₂SO₄ and concentrated under reduced pressure. The crude product was purified by column chromatography on silica (n-pentane/Et₂O = 10:1), and 10b was obtained as a colorless liquid (210 mg, 96%, 97:3 dr). (4S,5S)-5-Methyl-8-(tosyloxy)octan-4-yl acetate: $\left[\alpha\right]_{D}^{20} = -17.9^{\circ}$ (c 1.30, CHCl₃); ¹H NMR (400 MHz, $CDCl_3$): δ 7.77 (d, J = 8.4 Hz, 2H), 7.33 (d, J = 8.4 Hz, 2H), 4.79 (dt, J = 8.8, 4.0 Hz, 1H), 3.94-4.03 (m, 2H), 2.44 (s, 3H), 2.01 (s, 3H), 1.07-1.74 (m, 9H), 0.88 (t, J = 7.3 Hz, 3H), 0.83 (d, J = 6.9 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃): δ 170.8, 144.6, 133.1, 129.7 (2C), 127.8 (2C), 76.3, 70.6, 35.8, 33.2, 28.6, 26.6, 21.5, 21.0, 18.9, 14.2, 13.8; IR 2960, 2874, 1731, 1598, 1462, 1360, 1244, 1188, 1177, 1120, 1097, 1020, 964, 917, 815, 664 cm⁻¹; HRMS (ESI): m/z calcd for $C_{18}H_{32}O_5NS \ [M + NH_4]^+ \ 374.19957$, found 374.20048, calcd for $C_{18}H_{28}O_5NaS [M + Na]^+$ 379.15497, found 379.15546. **10b**: $[\alpha]_D^{20} =$ -31.9 (c 1.03, CHCl₃); ¹H NMR (400 MHz, CDCl₃): δ 3.51 (dt, J =7.5, 4.3 Hz, 1H), 1.24–1.54 (m, 9H), 1.12–1.20 (m, 1H), 0.93 (t, J =7.2 Hz, 3H), 0.90 (t, J = 6.9 Hz, 3H), 0.86 (d, J = 6.9 Hz, 3H); 13 C NMR (100 MHz, CDCl₃): δ 74.9, 37.9, 36.7, 35.6, 20.4, 19.4, 14.3, 14.1, 13.5; IR 3374, 2958, 2931, 2872, 1465, 1378, 1286, 1220, 1148, 1114, 1062, 1012, 970 cm⁻¹; HRMS (ESI): m/z calcd for C_9H_{19} [M + H - H₂O]⁺ 127.14813, found 127.14775.

(45,55)-4-Methylnonan-5-ol 10c, Ferrugineol. The same procedure as above was applied to 9c (432 mg, 2.00 mmol). The crude products were purified by column chromatography on silica as above. (4S,5S)-4-Methyl-1-(tosyloxy)nonan-5-yl acetate (639 mg, 90%, 97:3 dr) and 10c (261 mg, 93%, 98:2 dr) were obtained as yellowish or colorless liquids, respectively. (4S,5S)-4-Methyl-1-(tosyloxy)nonan-5-yl acetate: $[\alpha]_D^{20} = -14.7$ (c 1.39, CHCl₃); ¹H NMR (400 MHz, CDCl₃): δ 7.77 (d, J = 8.4 Hz, 2H), 7.34 (d, J = 8.0 Hz, 2H), 4.78 (dt, J = 8.5, 4.4 Hz, 1H), 3.95-4.04 (m, 2H), 2.44 (s, 3H), 2.02(s, 3H), 1.08-1.73 (m, 12H), 0.87 (t, J = 7.3 Hz, 3H), 0.84 (d, J = 6.9Hz, 3H); 13 C NMR (100 MHz, CDCl₃): δ 170.8, 144.6, 133.2, 129.7 (2C), 127.8 (2C), 76.5, 70.6, 35.8, 30.8, 28.6, 27.9, 26.6, 22.5, 21.5, 21.0, 14.2, 13.9; IR 2958, 2870, 1732, 1598, 1461, 1363, 1242, 1177, 1097, 1020, 955, 890, 815, 765 cm⁻¹; HRMS (ESI) calcd for $C_{19}H_{30}O_5NaS [M + Na]^+$ 393.17062, found 393.17096. **10c**: $[\alpha]_D^{20} =$ -27.2 (c 1.02, CHCl₃); ¹H NMR (400 MHz, CDCl₃): δ 3.48 (dt, J =7.9, 4.1 Hz, 1H), 1.23–1.52 (m, 11H), 1.11–1.19 (m, 1H), 0.91 (t, J =7.1 Hz, 3H), 0.90 (t, J = 7.1 Hz, 3H), 0.86 (d, J = 6.9 Hz, 3H); 13 C NMR 75.1, 37.9, 35.6, 34.2, 28.5, 22.8, 20.4, 14.3, 14.0, 13.5; IR 3379, 2958, 2930, 2870, 1462, 1379, 1145, 1114, 1015, 974 cm⁻¹; HRMS (ESI) calcd for $C_{10}H_{21}$ [M + H - H_2O] + 141.16378, found 141.16351.

(55,65,*E*)-6,10-Dimethylundeca-8,10-dien-5-ol 11a. The same procedure to synthesize 7a was applied to 6c (256 mg, 1.68 mmol), using *n*PrMgCl (2.0 M in Et₂O, 1.68 mL, 2.00 equiv). The crude product was purified by column chromatography on silica (*n*-pentane/EtOAc = 250:1–100:1), and 11a (244 mg, 74%, 94:6 dr) was obtained as a colorless liquid. [α]_D²⁰ = -13.3 (*c* 1.23, CHCl₃); ¹H NMR (400 MHz, CDCl₃): δ 6.17 (d, *J* = 15.6 Hz, 1H), 5.61–5.69 (m, 1H), 4.87 (s, 2H), 3.54 (td, *J* = 6.1, 3.7 Hz, 1H), 2.27 (dt, *J* = 13.7, 6.6 Hz, 1H), 1.98–2.06 (m, 1H), 1.84 (s, 3H), 1.57–1.66 (m, 1H), 1.24–1.48 (m, 7H), 0.87–0.94 (m, 6H); ¹³C NMR (100 MHz, CDCl₃): δ 142.0, 134.2, 129.2, 114.4, 74.6, 38.5, 36.9, 34.2, 28.4, 22.7, 18.7, 14.0, 13.3; IR 3380, 3080, 2958, 2930, 2871, 1646, 1608, 1457, 1378, 1312, 1123, 1004, 966, 881 cm⁻¹; HRMS (ESI) calcd for C₁₃H₂₅O [M + H]⁺ 197.18999, found 197.18996.

(65,75,E)-7,11-Dimethyldodeca-9,11-dien-6-ol 11b. The same procedure to synthesize 7a was applied to 6c (158 mg, 1.04 mmol).

The crude product was purified by column chromatography on silica (n-pentane/EtOAc = 250:1), and 11b (178 mg, 81%, 96:4 dr) was obtained as a colorless liquid. [α] $_{\rm D}^{20}$ = -12.7 (c 1.13, CHCl $_{\rm 3}$); 1 H NMR (400 MHz, CDCl $_{\rm 3}$): δ 6.17 (d, J = 15.6 Hz, 1H), 5.61–5.68 (m, 1H), 4.87 (s, 2H), 3.53 (td, J = 6.1, 3.7 Hz, 1H), 2.23–2.30 (m, 1H), 2.02 (dt, J = 14.5, 7.5 Hz, 1H), 1.84 (s, 3H), 1.57–1.66 (m, 1H), 1.40–1.47 (m, 4H), 1.26–1.35 (m, 6H), 0.87–0.92 (m, 6H); 13 C NMR (100 MHz, CDCl $_{\rm 3}$): δ 141.9, 134.2, 129.2, 114.4, 74.6, 38.5, 36.9, 34.5, 31.9, 25.9, 22.6, 18.7, 14.0, 13.3; IR 3379, 2958, 2929, 2860, 1608, 1457, 1377, 1124, 1017, 965, 881 cm $^{-1}$; HRMS (ESI) calcd for C $_{\rm 14}$ H $_{\rm 27}$ O [M + H] $^{+}$ 211.20564, found 211.20571.

(4S,5S)-5-Butyl-4-methyldihydrofuran-2(3H)-one 12a, (-)-cis-Whisky Lactone. A solution of 11a (80.0 mg, 0.40 mmol) in MeOH (5 mL) was cooled to -78 °C, ozone enriched O2 was bubbled through the solution until it turned blue, following by stirring for 5 min, and air was bubbled through the solution until the blue color disappeared. The reaction was quenched by the addition of NaBH₄ (36.3 mg, 0.96 mmol, 2.40 equiv), stirred for 1 h while warming up to rt, and extracted with CH₂Cl₂ (3 × 5 mL), and the combined organic layers were dried over MgSO₄ and concentrated under reduced pressure. The crude product was solved in MeCN (1 mL), and solutions of CuOTf·4MeCN (7.54 mg, 0.02 mmol, 0.05 equiv), 2,2'-bipyridine (3.13 mg, 0.02 mmol, 0.05 equiv), TEMPO (3.13 mg, 0.02 mmol, 0.05 equiv), and 1-methylimidazole (3.29 mg, 0.04 mmol, 0.10 equiv) each in MeCN (1 mL) were added. The reaction mixture was degassed for 5 min, the atmosphere in the reaction vessel was replaced by O₂, and the reaction was stirred for 2 h at 50 °C, quenched by the addition of H₂O (1 mL), and extracted with CH_2Cl_2 (3 × 5 mL); the combined organic layers were dried over MgSO₄ and concentrated under reduced pressure. The crude product was purified by column chromatography on silica (n-pentane/Et₂O = 5:1), and 12a was obtained as a colorless liquid (50.0 mg, 73%, 98:2 dr). $[\alpha]_D^{20} = -59.2^{\circ}$ (c 1.18, CHCl₃); ¹H NMR (400 MHz, CDCl₃): δ 4.39-4.44 (m, 1H), 2.68 (dd, J = 17.0, 7.8 Hz, 1H), 2.53-2.61 (m, 1H), 2.18 (dd, J = 17.0, 4.0 Hz, 1H), 1.59-1.69 (m, 1H), 1.43-1.55(m, 2H), 1.29-1.40 (m, 3H), 1.00 (d, J = 7.1 Hz, 3H), 0.91 (t, J = 7Hz, 3H); 13 C NMR (100 MHz, CDCl₃): δ 176.8, 83.6, 37.5, 32.9, 29.5, 28.0, 22.4, 13.8, 13.8; IR 3523, 2958, 2873, 1774, 1465, 1422, 1383, 1336, 1293, 1209, 1170, 1092, 1080, 994, 974, 940, 928 cm⁻¹; HRMS (ESI) m/z calcd for $C_9H_{17}O_2[M + H]^+$ 157.12231, found 157.12216.

(45,55)-4-Methyl-5-pentyldihydrofuran-2(3*H*)-one 12b, (–)-*cis*-Cognac Lactone. The same procedure as above was applied to 11b (127 mg, 0.60 mmol). The crude product was purified by column chromatography on silica as above, and 12b (78.8 mg, 77%, 98:2 dr) was obtained as a colorless liquid. [α]₀²⁰ = -64.8 (c 1.23, CHCl₃); ¹H NMR (400 MHz, CDCl₃): δ 4.41 (ddd, J = 9.4, 5.4, 4.4 Hz, 1H), 2.67 (dd, J = 17.0, 7.8 Hz, 1H), 2.52–2.60 (m, 1H), 2.17 (dd, J = 17.0, 4.0 Hz, 1H), 1.63 (q, J = 9.3 Hz, 1H), 1.45–1.53 (m, 2H), 1.26–1.39 (m, 5H), 0.99 (d, J = 7.3 Hz, 3H), 0.85–0.91 (m, 3H); ¹³C NMR (100 MHz, CDCl₃): δ 176.8, 83.6, 37.5, 32.9, 31.5, 29.8, 25.5, 22.4, 13.9, 13.7; IR 2956, 2865, 1779, 1462, 1423, 1383, 1335, 1294, 1212, 1168, 1079, 1006, 978, 932 cm⁻¹; HRMS (ESI) calcd for C₁₀H₁₉O₂ [M + H]⁺ 171.13796, found 171.13783.

(2S,6S)-2,6-Dichloroheptane-1,7-diol 14. To a solution of freshly distilled 1,7-heptandiol 13 (512 mg, 4.00 mmol) in MeCN (40 mL) were added CuOTf-4MeCN (149 mg, 0.30 mmol, 0.075 equiv), 2,2'-bipyridine (46.8 mg, 0.30 mmol, 0.075 equiv), TEMPO (46.8 mg, 0.30 mmol, 0.075 equiv), and 1-methylimidazole (49.7 mg, 0.60 mmol, 0.15 equiv). The reaction mixture was degassed for 5 min, and the atmosphere in the reaction vessel was replaced by O2. The reaction mixture was stirred for 3 h at rt, and (2R,5S)-2-(tert-butyl)-3,5-dimethylimidazolidin-4-one-TFA-salt (457 mg, 1.60 mmol, 0.40 equiv) and NCS (1.33 g, 10.0 mmol, 2.50 equiv) were added successively. The reaction was stirred for 18 h at rt, cooled to 0 °C, and diluted with EtOH (40 mL). After successive addition of NaBH₄ (756 mg, 20 mmol, 5.00 equiv) the reaction was stirred for 30 min at 0 °C, quenched by the addition of a saturated aqueous solution of NH_4Cl (20 mL), and extracted with EtOAc (3 × 60 mL). The combined organic layers were dried over MgSO₄ and concentrated under reduced pressure. The crude product was purified by column

chromatography on silica (*n*-pentane/Et₂O = 1:5), and 14 was obtained as a white solid (547 mg, 68%, 95:5 dr, >99:1 er). After recrystallization from CHCl₃, the dr can be enriched to >99:1. Mp: 79.3 °C; $[\alpha]_D^{20}$ = +51.1° (*c* 1.18, CHCl₃); ¹H NMR (400 MHz, CDCl₃): δ 4.00–4.06 (m, 2H), 3.78–3.83 (m, 2H), 3.68–3.73 (m, 2H), 1.90 (br. s, 2H), 1.76–1.86 (m, 4H) 1.67–1.76 (m, 2H); ¹³C NMR (100 MHz, CDCl₃): δ 66.9 (2C), 64.6 (2C), 33.6 (2C), 23.3; IR 3324, 2962, 2866, 1455, 1391, 1286, 1073, 1020, 998, 745, 611, 568, 510, 455 cm⁻¹; HRMS (ESI): m/z calcd for C₇H₁₅O₂Cl₂ [M + H]⁺ 201.04436, found 201.04433, calcd for C₇H₁₅O₂Cl³⁷Cl [M + H]⁺ 205.03846, found 205.03825; GC (Hydrodex-β-TBDAC, 170 °C isotherm, 1.1 mL/min He, 50:1 split) $T_{(2R,6R)}$ = 60.6 min, $T_{(meso)}$ = 63.6 min, $T_{(2R,6S)}$ = 70.5 min, isomeric ratio = 0:1:351.

1,3-Di-((R)-oxiran-2-yl)propane 15. To a solution of **14** (1.60 g, 7.69 mmol) in CH₂Cl₂ (27 mL) was added an aqueous solution of NaOH (1 M, 100 mL). After the reaction was stirred for 15 h at rt, the layers were separated and the aqueous layer was extracted with Et₂O (3 × 100 mL). The combined organic layers were dried over MgSO₄ and concentrated under reduced pressure. The crude product was purified by column chromatography on silica (Et₂O), and **15** was obtained as a colorless liquid (937 mg, 92%, 99:1 dr). $[\alpha]_D^{20} = +23.0$ (α 1.1, EtOH); ¹H NMR (400 MHz, CDCl₃): α 2.89–2.95 (m, 2H), 2.71–2.79 (m, 2H), 2.48 (dd, α 2 = 5.0, 2.7 Hz, 2H), 1.62–1.68 (m, 4H), 1.52–1.60 (m, 2H); ¹³C NMR (100 MHz, CDCl₃): α 52.0 (2C), 47.0 (2C), 32.2 (2C), 22.5; IR 2985, 2929, 2862, 1410, 1258, 1133, 911, 831, 764, 512 cm⁻¹; HRMS (ESI): α calcd for C₇H₁₃O₂ [M + H]⁺ 129.09101, found 129.09072.

ASSOCIATED CONTENT

Supporting Information

General experimental, ¹H and ¹³C NMR spectra, GC chromatograms, and crystal data. This material is available free of charge via the Internet at http://pubs.acs.org.

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Notes

The authors declare no competing financial interest.

REFERENCES

- (1) Gallezot, P. Chem. Soc. Rev. 2012, 41, 1538-1558.
- (2) Sheldon, R. A. Catal. Today 2011, 167, 3-13.
- (3) Swift, K. A. Top. Catal. 2004, 27, 143-155.
- (4) Willot, M.; Radtke, L.; Könning, D.; Fröhlich, R.; Gessner, V. H.; Strohmann, C.; Christmann, M. Angew. Chem., Int. Ed. 2009, 48, 9105–9108.
- (5) de Freitas, M. C.; Vieira, C. G.; dos Santos, E. N.; Gusevskaya, E. V. ChemCatChem **2013**, 5, 1884–1890.
- (6) Schwab, W.; Fuchs, C.; Huang, F.-C. Eur. J. Lipid Sci. Technol. 2013, 115, 3-8.
- (7) Winter, P.; Swatschek, J.; Willot, M.; Radtke, L.; Olbrisch, T.; Schäfer, A.; Christmann, M. Chem. Commun. 2011, 47, 12200–12202.
- (8) Winter, P.; Vaxelaire, C.; Heinz, C.; Christmann, M. Chem. Commun. 2010, 47, 394–396.
- (9) Winter, P.; Hiller, W.; Christmann, M. Angew. Chem., Int. Ed. **2012**, *51*, 3396–3400.
- (10) Shibatomi, K. Synthesis 2010, 2679-2702.
- (11) Oestreich, M. Angew. Chem., Int. Ed. 2005, 44, 2324-2327.
- (12) Fadeyi, O. O.; Schulte, M. L.; Lindsley, C. W. Org. Lett. 2010, 12, 3276–3278.
- (13) Britton, R.; Kang, B. Nat. Prod. Rep. 2013, 30, 227-236.
- (14) Bergeron-Brlek, M.; Teoh, T.; Britton, R. Org. Lett. 2013, 15, 3554-3557.
- (15) Dhand, V.; Draper, J. A.; Moore, J.; Britton, R. Org. Lett. 2013, 15, 1914–1917.

- (16) Holmes, M. T.; Britton, R. Chem.—Eur. J. 2013, 19, 12649—12652.
- (17) Vaxelaire, C.; Winter, P.; Christmann, M. Angew. Chem., Int. Ed. **2011**, 50, 3605–3607.
- (18) Hoover, J. M.; Stahl, S. S. J. Am. Chem. Soc. **2011**, 133, 16901–16910
- (19) For a review on aerobic Cu-catalyzed reactions, see: Allen, S. E.; Walvoord, R. R.; Padilla-Salinas, R.; Kozlowski, M. C. *Chem. Rev.* **2013**, 113. 6234–6458.
- (20) Semmelhack, M. F.; Schmid, C. R.; Cortes, D. A.; Chou, C. S. J. Am. Chem. Soc. 1984, 106, 3374–3376.
- (21) Markó, I. E.; Gautier, A.; Dumeunier, R.; Doda, K.; Philippart, F.; Brown, S. M.; Urch, C. J. Angew. Chem., Int. Ed. 2004, 43, 1588–1591.
- (22) Gamez, P.; Arends, I. W. C. E.; Sheldon, R. A.; Reedijk, J. Adv. Synth. Catal. (Advanced Synthesis & Catalysis) 2004, 346, 805–811.
- (23) Kumpulainen, E. T. T.; Koskinen, A. M. P. Chem.—Eur. J. 2009, 15, 10901–10911.
- (24) Pellissier, H. Tetrahedron 2013, 69, 7171-7210.
- (25) Könning, D.; Hiller, W.; Christmann, M. Org. Lett. 2012, 14, 5258–5261.
- (26) Ho, X.-H.; Oh, H.-J.; Jang, H.-Y. Eur. J. Org. Chem. 2012, 5655–5659.
- (27) Shibatomi, K.; Narayama, A. Asian J. Org. Chem. 2013, 2, 812–823.
- (28) Halland, N.; Braunton, A.; Bachmann, S.; Marigo, M.; Jørgensen, K. A. *J. Am. Chem. Soc.* **2004**, *126*, 4790–4791.
- (29) Halland, N.; Alstrup Lie, M.; Kjærsgaard, A.; Marigo, M.; Schiøtt, B.; Jørgensen, K. A. Chem.—Eur. J. 2005, 11, 7083–7090.
- (30) Brochu, M. P.; Brown, S. P.; MacMillan, D. W. C. J. Am. Chem. Soc. 2004, 126, 4108–4109.
- (31) Amatore, M.; Beeson, T. D.; Brown, S. P.; MacMillan, D. W. C. Angew. Chem., Int. Ed. **2009**, 48, 5121–5124.
- (32) Graham, T. H.; Horning, B. D.; MacMillan, D. W. C.; Hughes, D. Org. Synth. 2011, 42-53.
- (33) Banfi, L.; Guanti, G. Synthesis 1993, 1029-1056.
- (34) Kaiser, R. The scent of orchids: Olfactory and chemical investigations; Elsevier Science Publishers: New York; Editiones Roche: Amsterdam, Basel, 1993.
- (35) Fink, M. J.; Schön, M.; Rudroff, F.; Schnürch, M.; Mihovilovic, M. D. ChemCatChem 2013, 5, 724–727.
- (36) Kim, A.; Sharma, S.; Kwak, J. H.; Kim, I. S. Bull. Korean Chem. Soc. 2013, 34, 75–78.
- (37) Yadav, J.; Rao, R.; Kumar, B.; Somaiah, R.; Ravindar, K.; Reddy, B.; Khazim Al Ghamdi, A. *Synthesis* **2011**, 3168–3172.
- (38) Wu, Y.; Shen, X.; Tang, C.-J.; Chen, Z.-L.; Hu, Q.; Shi, W. J. Org, Chem. 2002, 67, 3802–3810.
- (39) Brenna, E.; Dei Negri, C.; Fuganti, C.; Serra, S. *Tetrahedron: Asymmetry* **2001**, *12*, 1871–1879.
- (40) Watanabe, Y.; Shimada, N.; Anada, M.; Hashimoto, S. Tetrahedron: Asymmetry 2013, DOI: 10.1016/j.tetasy.2013.10.016.
- (41) Marshall, J. A.; Garofalo, A. W. J. Org. Chem. 1993, 58, 3675–3680.
- (42) Hoye, T. R.; Ryba, T. D. J. Am. Chem. Soc. 2005, 127, 8256-
- (43) Giblin-Davis, R. M.; Oehlschlager, A. C.; Perez, A.; Gries, G.; Gries, R.; Weissling, T. J.; Chinchilla, C. M.; Peña, J. E.; Hallett, R. H.; Pierce, H. D.; Gonzalez, L. M.; Pena, J. E. *The Florida Entomologist* 1996, 79, 153.
- (44) Mori, K.; Murata, N. Liebigs Ann. 1995, 1995, 697-698.
- (45) Odriozola, J. M.; García, J. M.; González, A.; Gil, P. *Tetrahedron: Asymmetry* **1999**, *10*, 4627–4632.
- (46) Yadav, J.; Rao, C.; Prasad, A.; Khazim Al Ghamdi, A. Synthesis 2011, 2011, 3894–3898.
- (47) Ortuño, R. M.; Merce, R.; Font, J. *Tetrahedron* **1987**, 43, 4497–4506.
- (48) Benedetti, F.; Forzato, C.; Nitti, P.; Pitacco, G.; Valentin, E.; Vicario, M. *Tetrahedron: Asymmetry* **2001**, *12*, 505–511.

- (49) Ozeki, M.; Hashimoto, D.; Nishide, K.; Kajimoto, T.; Node, M. *Tetrahedron: Asymmetry* **2005**, *16*, 1663–1671.
- (50) Braukmüller, S.; Brückner, R. Eur. J. Org. Chem. 2006, 2006, 2110–2118.
- (51) Adrio, L. A.; Hii, K. K. M. Eur. J. Org. Chem. 2011, 2011, 1852–1857.
- (52) Devalankar, D. A.; Chouthaiwale, P. V.; Sudalai, A. *Tetrahedron:* Asymmetry **2012**, 23, 240–244.
- (53) Takahata, H.; Uchida, Y.; Momose, T. J. Org. Chem. 1995, 60, 5628-5633.
- (54) Armstrong, A.; Ashraff, C.; Chung, H.; Murtagh, L. Tetrahedron 2009, 65, 4490–4504.
- (55) Jiang, X.; Fu, C.; Ma, S. Eur. J. Org. Chem. 2010, 2010, 687-693.
- (56) Mao, B.; Geurts, K.; Fañanás-Mastral, M.; van Zijl, A. W.; Fletcher, S. P.; Minnaard, A. J.; Feringa, B. L. *Org. Lett.* **2011**, *13*, 948–951
- (57) Bénéteau, R.; Lebreton, J.; Dénès, F. Chem.—Asian J. 2012, 7, 1516–1520.
- (58) Maga, J. A. Food Reviews International 1996, 12, 105-130.
- (59) Poss, C. S.; Schreiber, S. L. Acc. Chem. Res. 1994, 27, 9-17.
- (60) Green, N. J.; Sherburn, M. S. Aust. J. Chem. 2013, 66, 267-283.
- (61) Hartmann, E.; Oestreich, M. Org. Lett. 2012, 14, 2406-2409.
- (62) Florence, G. J.; Cadou, R. Tetrahedron Lett. 2008, 49, 6784-6786.
- (63) Chow, S.; Koenig, W. A.; Kitching, W. Eur. J. Org. Chem. 2004, 1198–1201.
- (64) For a promising approach, see: Berkessel, A.; Günther, T.; Wang, Q.; Neudörfl, J.-M. Angew. Chem., Int. Ed. 2013, 52, 8467–8471
- (65) Tokunaga, M.; Larrow, J. F.; Kakiuchi, F.; Jacobsen, E. N. Science **1997**, 277, 936–938.
- (66) Bredihhina, J.; Villo, P.; Andersons, K.; Toom, L.; Vares, L. J. Org. Chem. 2013, 78, 2379–2385.
- (67) Li, X.; Burrell, C. E.; Staples, R. J.; Borhan, B. J. Am. Chem. Soc. **2012**, 134, 9026–9029.
- (68) Vigneron, J.; Dhaenens, M.; Horeau, A. Tetrahedron 1973, 29, 1055-1059.